



• Arnold Magnetic Technologies operates 10 factories: 2 in Europe, 6 in the USA and 2 in China.



• Arnold manufactures magnetic materials and produces sophisticated assemblies utilizing magnetic materials.



- Due to time constraints we'll limit ourselves to these three "elements" of the magnetic materials industry.
- We'll identify those characteristics that make a material useful.
- We'll examine some of the more important economic issues that have occurred over the past few years.
- Finally, we'll attempt to identify from which elements the next great magnet will be made and the research projects on both existing materials and the hunt for new materials.



- The single largest driver affecting the magnetic materials market is production and use of energy.
- Energy to produce other material goods, food, perform services, for creature comfort, transportation and for entertainment.
- Our very survival is dependent upon adequate supply of affordable energy.



- In the global community, the IEA provides production and consumption statistics.
- The OECD is comprised of the countries listed to the right on this slide shown in alphabetic order.



- Each year, an update is published showing energy production and consumption statistics.
- This slide shows total energy supply by fuel.
- An inflection point is evident at approximately year 2003.
- It is clear that fossil fuels represent the greatest fraction of energy sourcing.



• Although overall energy production is increasing, crude oil output is growing very slowly, predominantly in the Middle East.



• Natural gas output is surging primarily due to "unconventional drilling" with increases in output occurring throughout the world.



• Coal production has most noticeable increased in China.



• Electric power generation is highly dependent upon fossil fuels, especially coal and natural gas.



• Production of electric energy by nuclear generating systems has declined for several years, most noticeably in the OECD.



• Hydro production, which has been stable in the OECD is growing in China and in Southeast Asia in general.

Coal/peat	TW/h	Natural gas	TW/h	Oil	T₩h
People's Rep. of China	3 723	United States	1 045	Japan	153
United States	1 875	Russian Federation	519	Saudi Arabia	142
India	715	Japan	374	Islamic Rep. of Iran	67
Japan	281	Islamic Rep. of Iran	160	Mexico	48
Germany	272	Mexico	156	Indonesia	42
South Africa	243	United Kingdom	147	United States	40
Korea	225	Italy	145	Kuwait	36
Australia	173	Egypt	117	Pakistan	34
Russian Federation	164	Korea	116	Russian Federation	27
Poland	141	India	109	Egypt	25
Rest of the world	1 332	Rest of the world	1 964	Rest of the world	444
World	9 1 4 4	World	4 852	World	1 058

- In these summary tables, we see that Coal produces more than three times the electric output of natural gas.
- Oil is not generally used to produce electricity as it is reserved for transportation where it is far more useful.
- Growth of natural gas output in the US from unconventional drilling has allowed increased use of this fuel for electric production.
- The lesson we learn here is that each country will use the fuel that represents the best combination of cost effectiveness.



- The DOE has divided fuel type into non-renewable and renewable.
- (The illustration was from the DOE site c.2011).



- Renewable sources of electric generation were 13% of the total in 2011.
- As expected, oil (petroleum) was less than 1%.



- This is a powerful graphic!
- In 2011, output from non-hydro renewable energy sources was 4.8% of total.
- This forecast chart suggests that output will slightly more than double by 2035.
- If overall demand remains constant, renewables will account for about 11% of energy output.
- If overall electric energy demand increases at a CAGR of 1%, then non-hydro renewables will represent only 8.5% of the total.
- With hydro output remaining constant, total renewables (hydro plus non-hydro) will represent approximately 14.9% of electric production which means that 85% will still be produced by fossil fuels.



- Lawrence Livermore National Labs produces Sankey diagrams for over one hundred countries of the world showing sources of energy and output.
- For the US, it is notable that more than half of the energy is lost in production and distribution.
- Electric production is shown in the top tan-colored box.



• Sources of energy will continue to be an issue with at least these considerations.



- With the demand for electricity for commercial and personal use, it is apparent that we must strive for more efficiency in production, transmission and use.
- Magnetic materials play important roles in all three areas.



- While motors and generators can operate without permanent magnets, soft magnetic materials to direct the "flow" of the magnetic field are highly beneficial.
- On a weight basis, far more soft magnetic material is produced each year than permanent magnets greater than 20x more.



- I believe I am safe when saying that all magnetic materials are examined primarily for their hysteresis properties.
- In response to an externally applied field, magnetic materials generate an internal field an induced field.
- As we apply a stronger field in a direction (+H), weaken it and then apply it in the opposite direction (-H) we cause the internal field to respond accordingly.
- If we plot the applied field (H) versus the induced field (B), we have what is called the hysteresis loop of the magnetic material.
- We measure the combination of applied and induced field and represent it as the "Normal Hysteresis Loop".
- When we subtract the magnitude of the applied field from the combined field, the result is the "Intrinsic Hysteresis Loop".
- As we cycle the magnet around the hysteresis loop (by changing the applied field), we see that the 1<sup>st</sup> and 3<sup>rd</sup> quadrant are identical, though of reverse sign just as the 2<sup>nd</sup> and 4<sup>th</sup> quadrants are the same.
- The dashed lines in this graphic display initial magnetization.
- When initially magnetizing the sample, the terms Type 1 and Type 2 shown in this chart refer to the coercivity mechanism where Type 1 coercivity is due to a "nucleation" mechanism and Type 2 is caused by domain pinning.



- The magnetic hysteresis loop provides a great deal of information about the material.
- First, it identifies the type of magnet: soft or permanent.
- A magnetically "soft" material has very small HcB and the area within the loop is small, such as for the green curve and area shown here.
- On the other hand, a permanent magnet exhibits a much greater HcB and a larger area within the loop as shown in the tan portion of the figure.
- Both curves, as shown here, are Normal curves. The green soft material is typical of silicon iron (Si-Fe) and the red curve is typical of alnico 8.
- The values of Bs represent the magnetic saturation and Hs is the applied field strength at which saturation occurs.
- The energy to force a magnet around the loop is proportional to the area within the loop.
- Therefore, transformers and laminations in motors benefit from being made of soft materials with very low HcB and correspondingly small areas within the loop.

ARNOLD MAGNETIC TECHNOLOGIES							
	Definition and Examples						
			Н <sub>сВ</sub>	H <sub>cJ</sub>	H at Bsat	(~Msat.) <sup>A</sup>	
	ew	Soft	< 25	< 25	< 700		
	, Vi	Semi-Hard	25 to ~700	25 to ~700	~700 to 2000	X	
	Ove	Permanent	> 700	> 700	> 2000		
Г		Supermelley	0.002		2	7 900	
	Soft	Supermanoy	0.003		2	7,800	
		SiEo	0.08		100	10,000	
L		514 6	0.0		100	13,300	
Г	q	FeCrCo <sup>B</sup>	50-300		700	9,000 - 12,000	
	lan	Remalloy	250		1,000	15,200	
	÷	Vicalloy	250		1,000	14,500	
	Sen	Cunife	500		1,500	8,400	
	0)	Cunico	700		1,500	8,800	
Г		Alnico 5-7	740	750	2,000	13,500	
	'nt	Alnico 8	1,600	1,800	4,500	10,500	
	ane	Ferrite (Ceramic 8	3,000	3,500	6,000	4,500	
	Ĕ	SmCo 1:5 (R20)	8,700	30,000	30,000	10,000	
	Pel	SmCo 2:17 (R26)	9,900	25,000	45,000	11,500	
	_	NdFeB (N38SH)	12,000	20,000	35,000	14,000	
P	Soft and FeCrCo c	Semi-Hard materials us can be heat treated to yi	e Bsat.; PM mat eld many combi	erials use Msat. nations of Hc and	Bsat.		
Touch	es Your Wori	ld Every Day	© Arnold Magne	etic Technologies			

- There is actually a continuum of materials from very soft to very hard.
- We can divide them into soft and hard magnetic materials or we can split them into three categories (as shown here) wherein each category has specific benefits.
- For convenience we have assigned arbitrary values to distinguish one category from the next, but there is no hard rule regarding the differentiation in category.
- Note that alnico is considered (for the most part) a permanent magnet material, yet the Normal curve is used to describe magnetic properties, not the Intrinsic curve. (Alnico was the first permanent magnet with appreciable coercivity).
- When Ferrite magnets were invented, suddenly we had a material with HcJ much larger than HcB and it became common practice to provide both the Normal and the Intrinsic curves.
- Some materials, such as FeCrCo, can have various levels of HcB by adjusting the final heat treatment time, temperature and quench rate.



- The four most important characteristics of a soft magnetic material, determined by where the material is used, are shown at the top of this list.
- Electrical resistance affects eddy current and coercivity affects hysteresis loss of the material, both of which cause the material to absorb energy and heat up.
- It is the nature of the materials that most (but not all) magnetically soft materials are also physically "soft". That is they are malleable can be rolled to thinner strips or bent to shapes.
- If the material is deformed or mechanically worked, it may be desirable to thermally anneal it to re-establish low HcB.



- The magnetic properties of interest for soft magnetic materials lie in the entire hysteresis loop or in the 1<sup>st</sup> quadrant.
- The two most interesting values in the 1<sup>st</sup> quadrant are the maximum permeability which is the slope of a line from the origin to the "knee" of the curve such as the dashed line in this chart hitting the Supermendur curve at the mark.
- The higher the slope, the easier the material is to magnetize.
- The second figure of interest is the saturation magnetization on these curves, the value of Bs, and shown here for Deltamax with the near-horizontal dashed line.



- The most commonly used soft magnetic material is silicon-iron (Si-Fe, aka electrical steel) and as you may be aware, there are two major uses for it: Transformers/Inductors and Motors/Generators.
- Transformers are used not only for Electrical Power Distribution, but in virtually every power supply in every electrically powered electronic device in offices and homes.
- Motors have moved from industrial uses in the 1800's and early 1900's to home and automotive uses. In homes for example, we use motors in garage door openers, washing machines and dryers, furnaces for home heating, refrigerator compressor pumps, bathroom exhaust fans, garbage disposals, and electric powered shop tools.
- Motor usage in automobiles has expanded to include windshield wiper motors, seat adjusting motors, starters, cooling fans, passenger compartment fans, alternators, etc. Note however, many of these use low carbon steel cans for low cost and Si-Fe laminations are reserved for higher output, higher efficiency motors/generators.
- The newest example of lamination steel use in cars is the electric traction drive motor of hybrids or full electric vehicles.



• Now a look at permanent magnets.



- Permanent magnets have a list of key characteristics that differs from soft magnetic materials.
- For each application a subset of these characteristics will determine how well a material is suited to the application.
- All of these should be considered when selecting a material and when performing R&D.
- The most discussed figure of merit is "maximum energy product", abbreviated BHmax.
- A spring can be thought of as analogous to a magnet. We compress and release a spring to take advantage of its ability to store energy. Similarly, we magnetically "compress" the field from a magnet.
- The amount of energy we can temporarily store in a magnetic circuit is proportional to the magnet's maximum energy product. Since these are supposed to be "permanent" magnets, we would not want them to demagnetize, so also important is the value of HcJ which is a measure of resistance to demagnetization.
- Magnetic properties change with temperature, so it is important when selecting a material for an application to know what the properties are over the entire range of expected temperatures.



- For permanent magnets, we are primarily interested in the 2<sup>nd</sup> quadrant.
- This illustration is typical of the "demag" curves presented in product literature for ferrite, SmCo and Neo magnets.
- The key figures of merit for permanent magnet materials are indicated on this chart.
- The maximum energy product can be estimated from just the Br as shown in the equation assuming an appropriate value for recoil permeability.
- Conversely, the Br can be estimated when the maximum energy product is known.
- These calculations can only be made with a "straight line" material. (This does not work for alnico).
- As shown here, this material would be considered a straight line (Normal curve) or square loop (Intrinsic curve) material since the Normal curve is straight (at least to the maximum energy point).

ARNOLD' MAGNETIC TECHNOLOGIES	Material	First Reported	BH(max)	Нсі
	Remalloy	1931	1.1	230
Permanent Magnet	Alnico	1931	1.4	490
Development Timeline	PtCo	1936	7.5	4,300
p	Cunife	1937	1.8	590
Permanent Magnets have	Cunico	1938	1.0	450
been developed to achieve	Alnico, field treated	1938	5.5	640
<ul> <li>Higher Br and Energy Product</li> </ul>	Vicalloy	1940	3.0	450
(BHmax)	Alnico, DG	1948	6.5	680
<ul> <li>Greater resistance to</li> </ul>	Ferrite, isotropic	1952	1.0	1,800
demagnetization (Hci)	Ferrite, anisotropic	1954	3.6	2,200
<ul> <li>Most are still in production</li> </ul>	Lodex <sup>®</sup>	1955	3.5	940
<ul> <li>Exceptions</li> </ul>	Alnico 9	1956	9.2	1,500
Lodex was discontinued due to	RECo <sub>5</sub>	1966	16.0	20,000
use of hazardous materials in production and in the product	RECo <sub>5</sub>	1970	19.0	25,000
Cunife has been replaced by	RE <sub>2</sub> (Co,Fe,Zr,Cu) <sub>1</sub>	1976	32.0	25,000
FeCrCo	RE <sub>2</sub> TM <sub>14</sub> B	1984 2010	∫ 26.0	25,000
			35.0	11,000
Table based on information in <i>Advances in Dermanant Magnetian</i>	RE <sub>2</sub> TM <sub>14</sub> B		∫ 30.0	35,000
by Rollin J. Parker, p.331-332			52.0	11,000
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- What materials were used in motors prior to the discovery of the rare earth magnets?
- Prior to the discovery of Alnico and similar alloys in the 1930's, the materials used fell into a category called cobalt steels having high remanence (Br), but very low coercivity values of 150 Oersteds being typical.
- During the 1900's great strides were made in the development of improved permanent magnets as shown in this table.
- The first greatly enhanced and widely used commercial permanent magnet was alnico starting in the late 1930's.
- This was followed by Ceramic (Hard Ferrite) magnets starting in the mid 1950s.
- Increased values of both maximum energy product (BHmax) and Hci, resistance to demagnetization, were made culminating with the rare earth magnets: SmCo and "Neo" magnets (RE<sub>2</sub>TM<sub>14</sub>B).



• Michael Coey graphically expresses the improvements to magnetic materials with soft materials achieving ever lower Hc and permanent magnetic materials exhibiting higher Hci.



- Focusing on Coercivity and Energy Density (Maximum Energy Product), Matthew Willard provides this chart.
- Permanent magnets get better toward the upper right while soft magnetic materials improve toward the bottom left.



- A key characteristic in selecting the best magnet is the temperature range of the application.
- We note here that both Neo and ferrite magnets have a more limited useful temperature range than SmCo and alnico.
- Neo is not naturally a high temperature magnet material we try to make it work at high temperatures by substituting dysprosium for some of the neodymium.
- Ferrite can be theoretically used to over 350 °C. However, even by 150 °C, it loses 25% of its flux output and so that is a practical limit for motor applications.






			Magnet <u>Producers</u>			
rials for and to, by far, the	ALNICO	China Atlas Magtech Chengdu Amoeba China Hope Magnet HPMG Shanghai Dao Ye Many others	Japan & Korea Pacific Metals	USA Arnold T&S PM Co.	<b>Europe</b> SG Magnets Ltd Magnetfabrik Bonn Magneti Ljubljana	
tion of raw mate ets. They are als or magnets.	FERRITE	Anshang Dekang BGRIMM DMEGC Dongyang Gelin Jiangmen >50 more	Hitachi TDK	Hitachi TDK	Magnetfabrik Bonn Magnetfabrik Schramburg Ugimag	
ninates product rmanent magne argest market f	SmCo	Arnold Chengdu Mag Mat'l TianHe Tiannu Group Yunsheng >30 more	Hitachi Shin-Etsu TDK	Arnold EEC	Arnold Magnetfabrik Bonn Magnetfabrik Schramburg Vacuumschmelze	
China totally don manufacture of pe	NdFeB*	Anhui Earth-Panda AT&M BGMT Ningbo Jinji San Huan Thinova Yantai Zhenghai Yunsheng	Daido Hitachi Shin-Etsu TDK	Hitachi	Magnetfabrik Bonn (not licensed) Magnetfabrik Schramburg Magneti Ljubljana (not licensed) Vacuumschmelze (Neorem) *Licensed To sell into the USA	
	)	>150 more				
Qu	r World Touches Yo	our World Every Day	© Arnold Magnetic Tee	chnologies	38	

- China has for several years dominated in the production of permanent magnets.
- It has become increasingly obvious that they also consume the "lion's share" of product.
- For example, they produce about 80% of the world's rare earth magnets and export only 10% to the USA as magnets. Far more enter the US as magnets within products.
- If similar statistics apply to Europe, then China retains about 80% of the magnets they produce for re-sale domestically.





• In addition to quotas and tariffs (export and import), currency exchange rates also affect pricing of materials.



- Rare earth materials have experienced price inflation and market disruption.
- FOB China pricing is shown here.
- Notes on the chart indicate the main price drivers.
- It is educational to note that when prices became too high, users of rare earth magnets designed away from them and are now slow to return.



World Touches Your World Every Day...

## Australian Resources:

addressing the rare earth balance issue

**Current prices are not supportive of sustainable supply, let alone new production:** Consolidation, and an increasing focus on the environmental impact of mining have seen operating costs in China increase. In response to falling prices China's largest mine announced production curtailments in December 2012. Western producers have also indicated there are likely to be delays to growth projects, and Lynas announced it would look to set a floor for prices. While we believe prices could continue to fall in the short-term, we expect light-rare earth basket prices to increase from US\$16/kg currently to US\$18/kg in 2014, US\$22/kg in 2015 and US\$27/kg in 2016.

From document of same name, J.P. Morgan, Australia Equity Research, 11 July 2013

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42



- This is my attempt to forecast relative magnet prices going forward based on costs over the past 9 years.
- It shows a slow continual uptick in SmCo magnet pricing.
- It also assumes that Neo magnet pricing will bottom out in early 2013 and rise slowly going forward and this appears to be happening.
- Dashed lines provide pessimistic and optimistic pricing as well as a likely middle value.



- Rare earths have not been the only materials to experience pricing cycles as we can see here with iron.
- Iron is a major constituent of electromagnetic devices on a weight basis so a doubling of cost can have a profound affect on product cost even though the cost per kilogram is not as great as for NdFeB magnets.



- And even induction and synchronous reluctance motors, which do not use permanent magnets, require field generation from coils of copper wire.
- These charts show copper price swings of 500%.



- Aluminum is the third most prevalent material in the crust of the earth.
- Yet it too experiences wide price swings as supply and demand get out of synch.
- It is difficult for the mining-processing industry to respond to rapid swings in market demand.



• Copper is only slightly more prevalent in the earth's crust than neodymium.



• Both copper and aluminum are obtained from very large mines.



ARNOLD' MAGNETIC TECHNOLOGIES									
Major PM Materials									
	2010				2015				
	tons	<u>%</u>	<u>\$\$\$</u>	<u>%</u>	<u>tons</u>	<u>%</u>	<u>\$\$\$</u>	<u>%</u>	
NdFeB	67,300	10.5%	5,700	65.1%	80,015	8.2%	9,305	62.6%	
SmCo	2,310	0.4%	270	3.1%	3,510	0.4%	540	3.6%	
Ferrite	567,000	88.2%	2,600	29.7%	881,120	90.7%	4,760	32.0%	
Alnico	5,555	0.9%	125	1.4%	6,440	0.7%	185	1.2%	
Other	540	0.1%	65	0.7%	580	0.1%	75	0.5%	
Totals	642,705	100.0%	8,760	100.0%	971,665	100.0%	14,865	100.0%	
Sales \$\$ are in millions									
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- These are the most common permanent magnet materials.
- While neo magnets now represent between 62 and 65% of magnet sales on a dollar basis, ferrite dominates on a weight basis at over 88%.
- With the improved grades of ferrite both weight and dollar percentages are expected to rise.
- Ferrite magnets are considerably lower cost than rare earth magnets and the raw materials are widely available.

Ferrite Magnet Applications	s (2010)		
Motors - Automotive	18%		
Motors - Appliances	13%		
Motors - HVAC	13% 7	0%	
Motors - Industrial & Commercial	12%		
Motors - All Other	5%		
Loudspeakers	9%		
Separation Equipment	5%		
Advertising & Promotional Products	5%		
Holding & Lifting	5%		
MRI	3%		
Relays & Switches	1%		
All Other - Miscellaneous	11%		

- The majority of ferrite applications are motors.
- (Loudspeakers are linear motors).



- Switching to rare earth magnets...
- Since 2010, major topics of interest have been pricing and availability of rare earths and rare earth magnets.
- The materials shown here comprise the family of Rare Earth magnets.
- Although SmCo magnets are superior for elevated temperature applications, the combination of greater material availability and historically lower cost has propelled Neo magnets into a dominant position.
- However, for Neo to perform successfully at elevated temperature requires substituting dysprosium for some of the neodymium.
- Of late, the supply of dysprosium has not been adequate resulting in high material prices and a continuing long-term shortage is likely.
- SmFeN is an excellent material except that 1) it decomposes at a fairly low temperature preventing consolidation to full density and 2) because it must be used as a bonded magnet, maximum energy product is limited by the dilution with a non-magnetic binder.

ARNOLD MAGNETIC TECHNOLOGIES							
Rare Earth Magnet Applications (2010)							
Motors, industrial, general auto, etc HDD, CD, DVD Electric Bicycles Transducers, Loudspeakers Magnetic Separation MRI Torque-coupled drives Sensors Generators Hysteresis Clutch Air conditioning compressors and fans Energy Storage Systems Wind Power Generators Gauges Magnetic Braking Relays and Switches Pipe Inspection Systems Hybrid & Electric Traction Drive Reprographics Wave Guides: TWT, Undulators, Wigglers Unidentified and All Other	$\begin{array}{c} 24.0\% \\ 16.3\% \\ 8.4\% \\ 8.1\% \\ 4.6\% \\ 3.9\% \\ 3.3\% \\ 3.1\% \\ 3.0\% \\ 2.8\% \\ 2.4\% \\ 2.3\% \\ 1.9\% \\ 1.5\% \\ 1.5\% \\ 1.5\% \\ 1.5\% \\ 1.5\% \\ 1.5\% \\ 0.9\% \\ 0.8\% \\ 0.6\% \\ 0.3\% \\ 6.6\% \end{array}$						
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• Motors, generators, loudspeakers, represent about 67% of rare earth magnet usage.



• The market for rare earth magnets is experiencing dramatic shifts with a few examples shown here.



- A market for permanent magnets is wind power generation of commercial electricity by large, tower-mounted generators.
- In 2008, the American Wind Energy Association (AWEA) forecast installations of wind per the plum-colored bars on this chart.
- I have plotted blue bars to show actual installations each year through 2012.
- Actual installations have exceeded forecast largely due to government incentives.
- In addition to the potential to use permanent magnet generators, there are two other large motors in the tower: 1) the blade pitch motor and 2) the nacelle rotation motor.



- Prior to 2005, China's wind power industry was close to non-existent.
- Since then it has grown rapidly and now cumulatively China has more wind power generation installed than any other country at 26.7% of global generation.
- In 2011 alone, China produced 43% of all installations world-wide with a substantial portion being Gen-4, permanent magnet generator type. New installations dropped to 26.9% of global installations in 2012.
- In the USA only 233 permanent magnet generators have been installed out of more than 33,000. In China the number is higher, but no more than 25% of those installed in 2010-2011.



- Wind power generator type selection is a function of factors such as installed cost, operating maintenance costs, ease of maintenance, efficiency of power output, low speed cut-in and high speed cut-out, reliability, on-land versus off-shore installation, etc.
- Size and weight of the nacelle become an increasingly important factor as the output increases above 5 MW. By 10 MW, the only currently feasible technology is suggested to be superconducting generation.



- Hybrid automobiles and full electric vehicles are becoming increasingly more common in the US and Europe
- However, the economy of much of the world is such that cars are financially out-of-reach for the majority of the population.
- The less expensive electric bike is providing a path of upward mobility (if you'll pardon the pun) throughout southeast Asia and India.
- Although the amount of magnet material per unit is small, the quantities of bicycles are large.
- And because the market is so diverse, it is difficult to assign an accurate average magnet weight per vehicle (80 grams is used here though estimates range from 60 for motor-boosted bicycles to over 350 grams for high power scooters).
- Intermediate dysprosium levels (~4-6.5%) are required in the motors for electric bikes.
- High dysprosium (~8-10%) is required for EV's due primarily to the higher temperature of the application coupled with localized high demagnetization fields.
- Rate of hybrid or EV adoption is in-part driven (bad pun again) by the cost of gasoline, incentives such as rebates on car purchases, fees on non-hybrid vehicles, express lane commuting advantages, government mandates, etc.
- The rate of EV adoption has failed to keep up with estimates in the US, but other geographic regions are adopting faster. China, for example is mandating rapid EV adoption to reduce urban pollution.



- Sales for 2012 were approximately 461,250 units.
- 2011 and 2012 represent watershed years in that EV and PHEV sales have just started.
- Change in consumer preferences are due in large part to price of gasoline and overall economic prosperity.



- EV sales as a percentage of overall sales in the USA is increasing.
- 2012 ended with EVs being over 3% of sales!
- Despite the smaller number of models available in 2013 (59) versus JD Power's 124 there is no doubt that we are at a turning point regarding use of hybrid and electric vehicles.



- Global car production has been tracked by OICA and is charted here from 1997 through 2012.
- A second order regression fit allows us to extrapolate to year 2020 when it's possible that 90 million automobiles might be manufactured and sold.
- If 10% of those are to be hybrid vehicles they might use 3,461 metric tons of neodymium and praseodymium oxide which represents ~13.4% of 2010 reported global supply.
- However, dysprosium usage would require just under half of global supply.



- One of the main uses for rare earth magnets, predominately neo, is in electronic devices such as hard disk drives, CDs and DVDs where the magnet is used for driving the spindle motor, for positioning the read/write head, and providing a clamping force (in some CDs and DVDs).
- Even though the amount used per drive is small, the huge quantity of devices requires large amounts of rare earth magnets.
- Importantly, these devices use very low or even no dysprosium.
- This market is being eroded by expansion of SSD drives especially in portable devices which has resulted in double digit drops in desktop computers for two years in a row.
- Continuing markets for HDDs are for servers and high end desktop systems.





• Common, commercially available permanent magnetic materials.



- La-Co additions to strontium-ferrite permanent magnets has resulted in a 20%+ increase in energy product and high coercivity.
- Considering the relatively low cost of ferrite, enhanced performance suggests an even greater market share going forward.



- The dysprosium diffusion process provides more efficient use of dysprosium.
- Users may chose to 1) keep dysprosium as existing levels and benefit from the higher coercivity or 2) reduce the dysprosium to meet minimum acceptable coercivity and benefit from higher energy product.



• While not a new technique, it was practiced starting in the mid 1980s, the current method optimizes the effect of dysprosium diffusion.



- Magnequench has presented about the use of minor dysprosium additions to melt spun (jet cast) neo powders for use in hot-pressed and die-upset magnets.
- This slide shows a comparison of the coercivity of magnets made via conventional powder metallurgy and those made by jet-casting and die-upsetting.



- Daido has been a long practitioner of the die upset magnet manufacturing process, primarily for the production of hollow, radially oriented cylinders.
- Presumably the addition of dysprosium will enhance their product properties accordingly.



- Arnold has improved the composition and processing of SmCo to produce a grade of 2:17 SmCo with exceptional properties and with high production yield.
- For application temperatures above 180 °C, this material is superior from both a cost and performance standpoint.



- Arnold has also introduced magnets with complex magnetic orientation built-in.
- The resulting field patterns have reduced flux leakage and enhanced coupling in torquecoupled assemblies by over 30%.
- The patent applied for process is also applicable to ferrite and neo magnets.



- Alnico exhibits values of Br comparable with neo magnets.
- The problem with alnico lies in low coercivity.
- Projects are underway to develop higher coercivity alnico while minimizing loss of Br.






- Looking at the price chart for magnetic materials, the highlighted region shows target price and energy for new materials development.
- Permanent magnet R&D is focused on one or two objectives: increasing magnetic output and/or reducing the product cost all while using readily available materials.



- A key consideration of new materials development is maximum performance which is dependent upon co-parallel orientation of magnetic domains.
- This illustrations helps one to differentiate among crystals, domains, and particles.



- There is considerable confusion regarding these terms.
- Let us define what is meant by anisotropic versus isotropic and oriented versus unoriented.
- Most grains of magnetic material have an "easy axis of magnetization". This means that the crystalline material magnetizes in one orientation only. An example is the ferrite crystal shown above. In technical jargon, this is referred to as "uniaxial crystalline anisotropy".
- If the grains of magnetic material are not oriented during the manufacture of the magnet, when the magnetic material is subsequently "charged" (magnetized), it will be weaker than it could potentially be. However, it may be magnetized in any direction.
- If the grains are oriented during manufacture, then the magnet will have a net magnetic field in only that orientation.
- For any material, if the anisotropic magnetic powder is well-aligned during manufacturing it will have the greatest possible magnetic output for that material type.
- Typical % of perfect orientation is about 96-97%.



- This SEM photomicrograph of bonded ferrite shows the particle morphology and alignment.
- Although the particles are not perfect hexagonal platelets, they are generally flat and aligned well, much like this New England shale stone wall.

ARNOLD MAGNETIC TECHNOLOGIES	igin	of the l	Fie	el	d -	- 2	2			
	Atomic number	Element Shell Sub-shell	K (1) 1s	25	L [2] 2p	35	м (3 Зр	) 3d	N (4) 745 4p 4d 44	Number of unpaired spins (circles)
	21	Scandium	1	‡	††† +++	† ÷	***	† 0	† +	1
	22	Titonium	† +	‡	***	1	*** +++	†† 80	÷	2
Filling of the 3d shell	23	Vanadium	† ‡	1	***	ŧ	***	***	1 .	3
doesn't proceed until	24	Chromium	ţ	1	***	1	***	*****	∳ o	5-1
	25	Manganese	† +	1	***	1	***	11111	1	5
Incomplete limiting of the 3d shell is key to	26	Iran	1	+	***	† +	***	*****	1	4
nara- and	27	Cobelt	1	Ť	***	t	***	*****	1	3
ferromagnetism	28	Nickel	1	+	***	ŧ	***	*****	1	2
	29	Copper	‡	1	+++ +++	+	***	*****	† 9	1 1 1/4
	30	Zinc	+	1	+++ +++	+	***	*****	† *	0
		in se	P	air ted	ing e trai	əf e isit	lectro ion n	on spin 1etal el	s	
Our World Touches Your World Every Day	©.	Arnold Magnetic Tech	nnolog	gies						79

- What is the origin of (electro-) magnetism?
- If we represent the electron spins, within an atom, by arrows in a tabular form, we can look at several elements at once, as in this table of elements 21 through 30.
- As electrons are added, along with protons, neutrons, etc to form higher atomic number elements, the natural order would be to fill the electron shells in sequence. However, the energy level of the 4s shell is lower than the 3d shell and after the element Argon (atomic number 18), electrons fill the 4s shell for potassium and calcium before starting to fill the 3d shell.
- Electron spin pair imbalance is presented in the last column.
- Electron spin imbalance alone is insufficient to cause strong ferromagnetism.
- Additionally, the dipole-dipole interaction in the crystallized material is too weak to be the source of ferromagnetism.



- Heisenberg, using the quantum theory, in 1928 explained that as atoms with partially filled electronic shells at large distances from each other move closer to one another their shells begin to overlap and quantum mechanical exchange forces arise between the incomplete shells. The corresponding energy appears in the mathematical formulation as an "exchange integral".
- When the exchange energy is positive, as it is for Fe, Co and Ni, ferromagnetic properties are exhibited. This occurs when the atomic spacing (a) is about 3-4 times the radius of the incomplete shell (r).
- Additionally, some combinations of otherwise weak magnetic materials have strong magnetic characteristics. An example is MnAlC and MnBi. Alloying increases the exchange interaction for manganese, moving it from a negative value to positive.



		A		C TEC	D'	OGIES												
F	Group 1 LA 1 1.8654				Per	riodi	c Tak	ole o	f the	Elen	nent	s - C	ompl	lete				18 VIIIA 2 4
1	Hydrogen tit el,1	2 IIA					В	ased on	table fro	m vertex	42.com		13 IIIA	14 IVA	15 VA	15 VIA	17 VILA	Helium YiliA
2	Li Lithium	Be Berytlium	Categ	Liquid pories Aul Medit ine Earth Medit	Solid	Synthetic Name Gas Halayans							Beron	Carbon	Nitrogen VA +12,0,4,5(12,0)	Oxygen VA 2	F Fluorine vite	Neor VEA
3	11 22.9896 Na Sodium Piej 311	12 24.300 Mg Magnesium Piej 312	3	ecilion Metals • Evelt Metals for Metals 4 IVB	6 VB	Ken metalt Metalout E VIB	7 VIB	8 VIII	9 VIII	10 VIII	11 18	12 118	13 XXXXX Al Aluminum	14 20.0651 Silicon	P Phosphorus	16 32.003 Sulfur VA	17 33.43 Cl Chlorine VII.	18 Argon
4	19 38.0803 K Potassium (245-611	20 41.876 Ca Calcium [M] 412 42	21 44.9999 SC Scandium (Mr) 341 442 +1	22 67.007 Ti Titanium [// 3/2 4/2 -12.3.4	23 50.9419 V Vanadium (Ar) 345.412 +2.3.4.5	24 91.9961 Cr Chromium [84] 345 411 -\$2.3.6	25 54.500 Mn Manganese (M) 365.42 +2.3.47	25 51.44 Fe Iron (M) 345 442 42.3	27 54.9032 Co Cobalt [4/] 3/7 4/2 42.5	28 SLEED Nickel (N) 348 442 42.3	29 KLSH Cu Copper (M) MID 411 -1.2	30 63.68 Zn Zinc (Ar) 3410 452	31 (8.72) Ga Gallium (Ar) 3r10 4s2 4s1	32 72.44 Ge Germanium (Ar) 3411 42 42 +2,4	As As Arsenic (#15415 4:2 4:0 -0.543	34 74.96 Selenium (M) 3471 452 454 -4.042	35 71.50 Br Bromine [kr] 3410 452 455 -1.541	36 Krypte (H) Sent 40
5	37 80.4676 Rb Rubidium	38 #7.42 Sr Strontium (0) 562	39 00.9099 Y Yttrium (10) 441 5c2 -3	40 91.224 Zr Zirconium (%) 442.562	41 92,9964 Nb Nioblum (V) 446 St 1 +1.5	42 19.34 Mo Molybdenum (10) 445 5x1	43 % Tc Technetium [0] 4d 5d2 +0.7	44 191.67 Ru Ruthenium [0] 40 Set 40	45 182.996 Rh Rhodium (0) 481 5c1 -1	46 106.02 Pd Palladium (0) 4110 - 52.4	47 W7.000 Ag Silver (0) 4tt0 St1	48 112.411 Cd Cadmium (0) 4110 5c2	49 114.010 In Indium (0)] 4410 502 501 *3	50 198.70 Sn Tin [97] 4410 562 562	51 121.76 Sb Antimony (52) 4410 562 560 +1.513	52 122.6 Te Tellurium (10) 4410 502 504 +6.6/2	53 128.994     Iodine  (0) 4410 5x2 5y5 +1.5.7/1	54 K Xeno (00) 400 55
	55 132.965 Cs Cesium [76] 611	66 137.327 Ba Barium [76] 652	Lanthanide Series	72 STR.49 Hf Hatnium [24] 4714 542 642	73 500.940 Ta Tantalum [24] 4714 545 612	74 WD.54 W Tungsten [Xe] 4714 564 652	75 106.387 Re Rhenium (24) 474 545 542 +4.57	75 190,20 Os Osmium (24) 4714 545 542	77 192297 Ir Iridium pej-414.547.642 +5.4	78 m.sm Pt Platinum (he) 414 Set Set s2.4	79 196.967 Au Gold (Ne) 414 5410 541	BO 200.51 Hg Mercury [Se] 414 Serio 662 +1.2	81 201.303 TI Thallium (1)	82 3073 Pb Lead (%) 692	Bi Bismuth Phylice5	84 200 Polonium [Ptj] 594 224	85 2H At Astatine (Mg) Sec	Rn Rado Philite
7	87 223 Fr Francium [Pin] 711 +1	88 226 Ra Radium (Pre) 712 *2	Actinide Series	104 265 Rf Ruthertordium	105 362 Db Dutnium V0 0	106 286 Sg Seaborglum VB 0	107 34 Bh Noteium viii	108 zm Hs Hassium VIII 0	109 200 Mt Meitoerium van	110 201 DS Damestactions VIII 0	111 20 Rg Romtgenkan	112 24 Cn Copernicium	113 sh Uut Uruntrian SA 0	114 20 Uuq Ununquadium NX 0	115 sh Uup thunpentium	116 28 Uuh Ununtesium Via 0	117 nh UUS Unanseptian	118 Uur Uhunoct
		Lanthankies	67 130.306 La Lanthanum (xe) Set 612 +0	68 348.336 Ce Cerium [Xe] 41 541 612 +0,4	69 нажа Рг Разеодужкая (20145 052 +)	60 14626 Nd Neodymium [Ne] 46 812 +1	61 10 Pm Promethium (191 45 652 +)	62 100.36 Sm Samarium (20) 46.02 42.0	63 191.566 Eu Europium pej 47 62 43	64 107.20 Gd Gadolinium (Se) 47 Set 612 -0	65 198.929 Tb Terbium [xe] 49 612 -0	65 NZ3 Dy Dyspresium (Nej 470 652 +3	67 164.50 Ho Kolmium (xe) ent 642 +3	68 167.29 Er Erbium [xa] 472.462 +3	69 162.00 Tm Thulium [20] 473 652 +3	70 173.04 Yb Ytterbium (14) 474.662 42,3	71 174.96 Lu Lutetium (%) 474.541.652 *)	
		Activides	Actinium [Pe] 641 752 +]	90 70.00 Th Thoeluss [Pa] 602 752 -4	Pa Protactinium	U Uranium	Neptunium	PU Plutonium	Am Americium	Cm Curium	Berkelium	Cf Californium	ES Einsteinium	Femium	Mcl Mendelevium	Nobelium	LT Lawrencium	
			/					-	-	~			2		-	1		82

- Let's work with the periodic table to see what elements are likely candidates for use in magnetic materials.
- I will use a method similar to that of Bill McCallum of Ames Laboratory who kindly shared his notes with me a year or two ago.
- And I should point out that this table was obtained from Vertex in Excel format. It has been modified to simplify the information in each cell. Go to www.vertex42.com for this and other useful spreadsheets and documents.
- This first table lists all of the elements... so let's start thinning the list.

	Group 1					P	eriod	ic Ta	able	of th	e Ele	emen	ts					1
1	1.00794							No	eventhet	ic olom	onte							2
	Hydrogen							INU :	synthet		ents		12			**	47	Hel
L	111 +1,-1	IIA											IIIA	IVA	VA	VIA	VIIA	Y
3	1.1	4 1.00210 Ro	Phase Gas	e at STP Liquid	Solid	Synthetic							6 14.011	6 12.0117	7 14,0067	8 13.9994	9 12,9904	10
	Lithium	Beryllium	Cate	gories									Beren	Carbon	Nitrogen	Oxygen	Fluorine	N
Ļ	(94) 211 +1	1966) 272 42	Alta	Alkali Metals Ine Earth Metals		Nalle Gas Hakigens							#A +3	10.8. 42,40-4	¥A +1,2,3,4,5(1,2,3	-2	-1	
1	Na	12 A.M	1n Pa	ecition Metals re Ewth Metals		Non-metals Metaloids							13 x.sen ΔΙ	14 Si Si	15 38.97% P	16 32.003	17 3.63	18
	Sodium	Magnesium	3	Poor Metars	5		7	8		10	11	12	Aluminum	Silicon	Phosphorus	Sulfur	Chlorine	A
	(Ne) 351 +1	[Nie] 312 +2	1118	IVB	VB	VIB	VIIB	VIII	VIII	VIII	IB	118	43	10X +2,4/4	+1,5/3	10A +4,5/2	¥88. +1,5,741	
1	K	Ca	Sc	Ti	23 90.941	Cr	Mn	Fe	Co	Ni	Cu	7n	Ga	Ge	As	Se	Br	3
	Potassium [Ar]-411	Calcium (Ar) 412	Scandium	Titanium (Ar) 362-462	Vanadium (Ar) 343-412	Chromium (Ar) 345 4c1	Manganese (M) 365 462	Iron (M) 346 462	Cobalt (M) SIT 4:2	Nickel (N) 348 442	Copper (Art 3410 4t1	Zinc [4:] 3:10 4:2	Gallium	Germanium [Ar] 3ml 42 42	Arsenic [Ar] 5410 442 440	Selenium (xr) 3rm 4/2 4/4	Bromine [87] 3610 462 465	Kry (Ar) Set
3	7 80.4676	38 #1.42	39 83.9019	40 91.224	41 \$2,9064	42 11.54	111	44 191.67	45 182.998	46 106.02	47 987.06	48 112.411	49 114.010	50 114.71	51 121.76	52 127.6	53 126.904	54
	Rubidium (V) Set	Strontium (N) Se2	Yttrium (6) 441 562	Zirconium (M) 442 5x2	Niobium (Kr) 444 St 1	Molybdenum (Kr) 445 Sr1	1 C	Ru Ruthenium (0) 40 Sc1	Rhodium (%) 440 Set	Pd Palladium (0) 410	Ag Silver poj 4/10 Set	Ccd Cadmium (Vr) 4r10 5c2	In Indium (N) 4410 5o2 5e1	5n Tin (9/) 4/10 552 5p2	SD Antimony (N) 4110 5c2 5p3	1 C Tellurium (Kr) 4210 502 5p4	lodine (%) 4410 512 5p5	Xer (94) 440
0	41 (5 132,908	66 137.327	*)	72 171.49	73 100.940	74 103.64	75 106.287	76 110.23	77 112.217	78 10.870	79 196.96	7 80 200.55	81 204.303	82 207.2	83 200.00	84 28	85 210	86
	Cs Cesium (Ne) 4x1	Ba Barium [Ne] 662	Lanthanide Series	Hf Hatnium (Ne) 4714 542 612	Ta Tantalum (Ne) 4714 545 652	W Tungsten (24) 414 544 652	Re Rhenium (24) 474 545 552	Os Osmium [Pe] 474 545 512	Ir Iridium pej 414 547 642	Platinum (Re) 4714 Set Set	Au Gold (Ne) 414 Settl Le	Hg Mercury 1 pej 414 Setto 662	TI Thallium Phil Set	Pb Lead Phil 492	Bi Bismuth Phyl Sep3	Po Polonium Ptgl 5p4	At Astatine Mg1 Re5	Ra Pa
80	17 223	88 225	-	104	-		-4,67	*1,*	-0,4	~~	*1,5		*1,5	42,4	40,5	-4,4	10	114
	Fr	Ra Radium	Actinide	RI	Db	sg	Bh	Hs	ALL .	Ds	Rg	Cn	Uut	Dug	Uup	Uuh	Uus	-0
	+1	ag foot use	Series							-			1.1					
			67 130.996	68 HE 116	69 HE 380	60 144.24	51 1 H	62 100.36	63 191.964	64 117.25	65 198.92	66 112.3	67 164.SI	68 167.299	69 Hill 104	70 173.04	71 174.967	7
		Lanthank	La Lanthanum (xe) Set esz	Ce Cerium [pe] an Sat Baz	Pr Praseodymium (Pe) 45 812	Nd Neodymium (No) 41 612	Pm	Samarium (24) 45 652	Eu Europium pej 47 62	Gd Gadolinium (Se) 47 Set 612	Tb Terbium (xe) 49 6s2	Dy Dyspresium [xe] 410 es2	Ho Holmium (ps) ant 6s2	Er Erbium (pa) 412 412	Tm Thulium pej-413-612	Yb Ytterbium (Nr) 474 662	LU Lutetium pej ana sen esz	
		_	89 227	90 232.030	91 231.03	92 231.629	10 D			11	10	1		100 10	121. 14	-	103	
		Actinides	AC Actinium	Th morium (Pa) 662 7a2	Pa Protactinium	Uranium	Np.	Pu	Am	Cm	Bk	Californitum	Es	Fm	Md	No	L.F.	

• This chart has eliminated the synthetic elements – those that are man-made and do not survive more than a very short time.

	Group 1					P	eriod	lic Ta	able	of the	e Ele	emen	ts					12
1	1.00794	1					No svr	thetic a	and no	adioact	tive ele	ments						2
	Hydrogen	,					NO Syr			aulouc		mento	12	44	- 45	16	47	Hel
	111 +1,1	IIA	122	02225									IIIA	IVA	VA	VIA	VIIA	1
3	Li	Be	Gas	e at STP Liquid	Solid	Synthetic							B	C	7 SCORD	8 13.9994 O	9 11.9904	10 N
	Lithium	Beryllium	Cate	pories		10.00 Que							Beren	Carbon	Nitrogen	Oxygen	Fluorine	Ne
	+1 1 20 0000	42	Alka	ive Earth Metals		Halogens							-1	42,444	+12,2,4,9(12,3	2	-1	
ľ	Na	Ma		re Earth Metals		Metaloida							AI	Si	P	S	CI	A
	Sodium Petan	Magnesium Piel 352	3	4	5		7	8	9	10	11	12	Aluminum	Silicon	Phosphorus	Sulfur	Chlorine	Arg
1	e1 9 39.0963	42 20 40.075	1118	IVB 22 47.367	VB 23 90.9419	VIB 24 \$1.9961	VIIB 25 34.530	VIII 26 25.045	VIII 27 94.9032	VIII 28 52.6834	18	118 130 63.489	43 31 48.722	42,4/4 32 72.64	43,5/3	=4,6/2 34 78.96	+1,5,7/-1 35 78.904	36
	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	K
	(AS 411	(Ar) es2	Scandium (A) 341 412	Titanium [A/] 3/2 4/2	Vanadium (Ar) 343 412	[Ar] 345 4r1	Manganese [M] 365 462	(Ar) 346-442	Cobalt (4) 347 452	Nickel (M) 348 462	Copper (Ar) 3410 4(1	Zine [4/] 3/10 4/2	Gallium [Ar] 3410 462 491	Germanium [Ar] 5r10 4:2 4:2	Arsenic [2/] 3/10 4/2 4/0 +1 5/1	Selenium (M) 3410 452 494	Bromine [3r] 3e10 4s2 4e5	Krys (A) 3410
3	7 30.4670	38	39 81.9009	40 91.224	41 52,9064	42 10.94		44 101.07	45 182.906	46 106.62	47 107.000	48 112.411	49 114.010	50 114.71	51 121.7	52 127.6	63 T28.904	54
	Rubidium	Strontium	Yttrium	Zireonium	Niobium	Molybdenum	I.C.	Ruthenium	Rhodium	Pd Palladium	Ag Silver	Cd Cadmium	In Indium	Sn Tin	Antimony	Tellurium	Iodine	Xen
	1+1	Kr  512 42	(00) 441 552 +3	(0) 442 512 +4	[N7] 464 St1 +3,5	(Kr) 445 511 +6		(00) 467 St 1 +3	(0) 485 St1 +3	(90) 4410 +2,4	(K) 4210 511 +1	[Kr] 4410 5x2 +2	(90) artii \$12 5y1 43	[Kr] 4410 (c.) 5p2 +2,4	(K)] 4410 512 5p3 +0,513	(K/) 4410 St2 Sp4 =4,6/2	[Kir] 4810 Sti2 Sp5 +1,5,7/1	pog entit
ľ		Ba		72 1748 Hf	73 100.940 Ta	W 101.04	Re Re	76 196.25 OS	17 192200	78 m.srs Pt	29 196.96 AU	80 2003 Ha	TI	82 30/2 Pb	Bi	Po	AT	R
l		Barium	Lanthanide	Hafnium	Tantalum	Tungsten	Rhenium	Osmium	Iridium	Platinum	Gold	Mercury	Thallium	Lead	Bismuth	Pulsalare	Artolina	1.0
ŀ	C125	eg Ivaliane	aeries	14	e2 fortiere property		+4,67 (Ve) 414 Sec 902	*3,4	*1,4	\$2,4	+1,3	1,2	91,3	1791 994 42,4	43'2 full dio	22	10.00	1
ľ		Ra		Rf	Db	Sg	Bh	Hs	Mit	Ds	Rg	Ctt	Unt	Uuq	Uup	Uuh	Ulis	U
l		- Manadalana	Actinide Series	Automation in the second	Substance -	-	100	Paratala -	and a second second		-	independent of the second seco	in a state of the second	the second s	in a second second	1000	-	
			SOM REAL		2					6				2.5				_
		des	67 131.906	68 140.1% C.e	59 148.980 Pr	Nd	Pitt	52 H0.36	63 191.964 Fu	Gd HZ M	65 HL 92	66 123	67 HO	68 167.99	59 162.53 Tm	70 171.04 Yh	71 174.907	1
		Lanthan	Lanthanum [Pe] Set 6s2	Cerium (Ne) 41 541 612	Praseodymium [76] 45 6s2	Neodymium [rie] 45 (c2	Presentations	Samarium Diej 45 652	Europium (24) 47 612	Gadolinium [Ne] 47 Set 442	Terbium [24] 45 612	Dysprosium pet ant 6s2	Holmium (Ne) 411 812	Erbium (Ne) 412 602	Thulium (te) 413.662	Ytterbium (24) 414 652	Lutetium (Ne) 414 Set 852	
			-1	-1,6			10 11	43	-0	-1		-	-1	<b>4</b>	-1	-01	103	
		thide		Th	Pa	Literations -	Np	Pti	Am	Cm	BX	CE	Es.	Em	Mc	No:	Lt	
		¥		10000				a serie and a series of		2000				1001-000				

- And let's eliminate the radioactive elements we won't want to put them in commercial magnets!
- As a side note Arnold has patents on Actinide magnets using uranium, iron and boron. They have interesting properties. But every known salt of uranium is either toxic or carcinogenic.

	Group 1 LA					P	eriod	lic Ta	able	of th	e Ele	emen	ts					v
1	L00794					1	lo synt	hetic, n	o radio	active, i	no inert	elemer	nts				1	1
	Hydrogen tel el,1	2 IIA 4 1.01212	Phas	a at STP									13 IIIA	14 IVA	15 VA 7 14.000	15 VIA	17 VILA	11
	Li Lithium pegan	Be Berytlium	Gas Categ	Liquid pories	Solid	Synthetic Nume Gas							Boron	C Carbon	Nitrogen	O Oxygen VA	F Fluorine VIX	
1	Na	12 21.300 Mg	Aks Tri Pa	ine Earth Metals ecution Metals re Earth Metals Foor Metals		Non-metals Metaloids							13 36.9800 Al	14 20.0000 Si	15 30.973	16 32.000 S	17 33.413 Cl	10 j
27 "	Sodium piej 3s1 +1	Magnesium piej 312 42	3 IIIB	4 IVB	6 VB	6 VIB	7 VIIB	8 VIII	9 VIII	10 VIII	11 18	12 IIB	Aluminum BA 43	Silicon NX 42,4/4	Phosphorus 10 41,563	Sulfur VA +4,8/2	Chlorine VIIX +1,5,761	3
1	) کاری کار الا الا الا الا الا الا الا الا الا ا	20 41.670 Ca Calcium (4/) 412 42	21 44.9999 SC Scandium [Ar] 341 442 -0	22 0.00 Ti Titanium [At] 342 442 42,3,4	23 50.9411 V Vanadium [Ar] 343 412 +2,3,4,5	24 11.9961 Cr Chromium [Ar] 345 411 ~2,3,6	25 54.500 Mn Manganese (M) 365 462 42,3,4,7	25 51.640 Fe Iron (M) 365 462 42.3	27 54.903 Cobalt [44] 547 462 423	28 SLIRO Nickel (N) 388 462 423	29 ELSH Cu Copper [Ar] 3410 411 -1,2	30 60.600 Zn Zinc (Ar) 5410 412 -2	31 08.723 Ga (Aallium (Ar) 3410 452 4pt «3	32 72.64 Ge Germanium [Ar] 3410 452 452 +2,4	33 74.92% Arsenic [Ar] 5419 462 463 +0,543	34 74.96 Selenium [M] 3471 452 454 -4,0/2	35 71.504 Br Bromine [14] 3610 462 465 +1,541	1.0
3	Rb Rubidium	38 #7.82 Strontium (99) 512 42	39 83.9999 Y Yttrium (0) 441 542 ~3	40 91.224 Zr Zirconium (0) 412 512 -4	41 \$2,996 Niobium (N) 446 \$11 -3,5	42 19.94 Mo Molybdenum (10) 445 5c1 45	TC Second	44 191.97 Ru Ruthenium [0] 40 Sc1 +3	45 182.99 Rh Rhodium (0) 48 5c1 +3	46 106.42 Pd Palladium (0) 4410 ~2,4	47 WZ.96 Ag Silver (0) 4410 Set +1	48 112.411 Cd Cadmium (0) 4410 5c2 42	49 milen In Indium pict acto Go2 Get <0	50 198.70 Sn Tin (kr) 4410 552 562 +2,4	51 121.76 Sb Antimony [N] 4410 522 540 +0,543	52 122.6 Te Tellurium (10) 4210 502 5p4 -6,6/2	53 126.994     lodine  (10) 4410 5x2 5p5 +1,5,7/1	
	Ca	66 037.327 Ba Barium [26] 652 42	Lanthanide Series	72 sm.ee Hf Hafnium [24] 4542 612	73 100.940 Ta Tantalum [76] 4714 545 652 -6	74 sec.se W Tungsten pej 414 544 652	76 106.207 Re Rhenium [24] 414 545 552 -6,57	76 196.20 Osmium [24] 474 545 542 +3,4	77 98.20 Ir Iridium pej 414 547 642 +3,4	78 m.an Pt Platinum (Pe) 414 Set Set <2,4	79 196.96 Au Gold (ne) 4714 5410 Ret +1,3	B0 200.50 Hg Mercury [Pe] 4714 Set2 662 +1,2	81 201303 TI Thallium (1)3	62 2022 Pb Lead (19) 692 <2,4	83 an.w Bi Bismuth (http://sea <3.5	Po. Polentian	At	
	Fr		Actinide	RI	Ob	sg	Bh	Hs	MI	Ds	Rg	Cn	Uut	Uuq	Uup	Uuh	Uus	
			20102															_
		Lanthankles	67 La Lanthanum (pe) Set 652 -7	Ce Cerium pej an Sat 612 -3,4	09 148.388 Pr Praseodymium [76] 45 452 -0	0 1424 Nd Neodymkum [Ne] 44482 -)	Pm	62 196.36 Samarium [26] 45 652 -42.3	63 101.36 Europium [26] 47 562 	Gd 102.25 Gd Gadolinium (Se) 47 Set 452 *)	65 194.95 Tb Terbium [29] 49 612 <)	Dysprosium (xe) = = = = = = = = = = = = = = = = = = =	Ho Holmium (xe) ant 6x2 -3	68 167.200 Erbium [76] 412 612 -3	59 164.00 Tm Thulium (20) 413 652 -)	Ytterbium pejeresiz 423	21 174367 Lu Lutetium (Pe) 474 Set 012 -0	
		Actinides		Th	Pa	U	Np	Pu	Am	Cm	Bk	Gf	Es	Fm	Md	No	Li	

• Let's also eliminate the noble gases – the inert elements. No magnetics available here.

•	Group 1 LA					P	eriod	lic Ta	able	of th	e Ele	emen	ts				1	v
	Н					NO SYI	nthetic,	no radi	oactive	, no ine	rt, no to	oxic ele	ments					
	tist e1,-1	2 IIA											13 IIIA	14 IVA	15 VA	16 VIA	17 VIIA	
3	Li	Re	Phase Gas	e at STP Liquid	Solid	Synthetic							6 M.211 B	6 12.0117 C	7 14.0067 N	8 13.9994 O	9 12.9904	10
	Lithium pegart		Cate	gories Abul Metals		Nuble Gat							Boron	Carbon	Nitrogen	Oxygen	Fluorine	
1	+1 1 22.9090	12 24.300	Aba	ine Earth Metals endion Metals		Hakiyest Non-metals							-1	+2,4/4	+12,3,4,5/12,3	-2	17 35.453	1
	Na Sodium	Mg Magnesium		Poor Metaris			,			10		12	Al	Silicon	P Phosphorus	S Sulfur	Cl	
41	piej 311 +1	Piel 312 42 20 #8478	IIIB	IVB	VB	VIB	VIIB	VIII	VIII	VIII	18	IIB	43 74 69 771	100 42,4/4 72 7244	41 41,543	VA +4,6/2 24 70.90	VIX. +1,5,741	_
1 1 1 1 1 1	K Potassium (Rej.411 +1	Ca Calcium	Sc Scandium (M) Set 4:2 -1	Ti Titanium (4) 342 442 42,3,4	V Vanadium (Ar) 343 412 +2,3,45	Cr Chromium (Ar) 345 4t1 +2,3,6	Mn Manganese [M] 36 42 43,47	Fe Iron (M) 365 462 423	Co Cobalt (M) 307 462 423	Ni Nickel (4) 348 442 423	Cu Copper (Ar) Sets 4ct -1,2	Zn Zinc (M1 Settle 4:2	Ga Gallium (Ar) 3410 452 461	Ge Germanium [Ar] 3rtl 4:2 4;2 +2;4	As Annale Annale	Se Selenium (M) 3411 462 464 -4,6/2	Bromine [14] 3610 462 465 -1,541	
3	Rubidium	38 #7.82 Strontium (%) 5x2 42	39 80.9999 Y Yttrium (0) 445 5c2 +3	40 91.29 Zr Zirconium (0) 402 562 -4	41 \$2,004 Niobium (0) 444 \$11 -3,5	42 16.54 Mo Molybdenum (V) 445 5c1 -6	TC beneficient ant ant ant	44 90.00 Ruthenium [0] 40 Set 43	45 182.99 Rh Rhodium (0) 48 5c1 +3	46 106.42 Pd Palladium (0) 4410 -2,4	47 WZ.96 Ag Silver (0) 4410 Set -1	Cd Cethnian	49 THERE In Indium (bi) 4410 552 5pt +3	50 198,71 Sn Tin (10) 4410 562 562 +2,4	Sb	Te Te Televise	53 128.994     lodine  00] 4410 5x2 5p5 +1,5,7/1	No.
	Ca	66 137.327 Ba Barium [56] 652 42	Lanthanide Series	72 STR.49 Hf Hatnium [24] 474 542 642 44	73 sause Ta Tantalum (24) 414 543 612 +6	V Tungsten pej 414 544 652	75 106.307 Re Rhenium [24] 474 545 552 +6,57	76 190.20 Osmium [24] 474 545 652 +3,4	77 98.20 17 17 17 17 17 17 17 17 17 17	78 m.an Platinum pej 414 sen sen <2,4	79 HK.SK Au Gold (Re) 4714 Sattle Rat -1,3	Hg	TI Durbum 75.40	Pb	83 30.00 Bismuth Ptgl 645 43,5	Po. Palantan nurs	At	Ran
1	Fr		Actinide	RI	Ob Indexed	Sg	Bh	Hs	ML ML materialism	Ds	Ro	Cn	Uut	Uuq	Uup	Uuh	Uus	
L			Series															_
		Lanthankes	67 536.396 La Lanthanum [Xe] 541 652 +7	68 148.116 Ce Cerium [xe] 40 541 612 -0,4	69 946.386 Pr Praseodysius [76] 43 612 <0	60 144.24 Nd [50] 444.02 -0	Pm	62 190.36 Samarium (26) 45 652 42.3	63 191.96 Eu Europium [26] 47 662 42,3	64 197.23 Gd Gadolinium (54) 47 541 642 49	65 198.92 Tb Terbium [24] 49 612 <)	0 65 162.1 Dy Dysprosium [Xe] 410 612 +0	67 164.92 Ho Holmium (xe) ent 642 +5	68 167.209 Er Erbium [70] 412 612 +3	69 168.304 Tm Thulium (20) 413.652 +3	70 state Yb Ytterbium pej-states 423	21 174.367 Lu Lutetium pej en 4 Set 012 -0	
		Activides	Ac	Th	Pa Pa	U	Np	Pu	Am	Cm	Bk	Gf	Es	Fm	Md	No	Lr Lr	

- Here we have eliminated radioactive elements.
- It's fair to say "no" to toxic materials so they are gone from this list.

1	Group 1 LA Laterse				No s	Po ynthetio	e <b>riod</b> c, no rad	lic Ta	<b>able</b> e, no ir	of the	e Ele toxic a	emen nd no ra	i <b>ts</b> are eler	nents			[	v
3	Hydrogen tet e1,1 E341 Li Lithium	2 IIA Be	Phase Gase	e at STP Liquid	Solid	Synthetic							13 IIIA 5 M.DIT B Beren	14 IVA 6 120187 C Carbon	15 VA 7 scoor N Nitrogen	16 VIA 8 13.999 Ozvata	17 VIIA 9 12.1954 Fluorine	1
1	piej 2st +1 22.90% Na Sodium piej 3st	12 24.300 Mg Magnesium Piej 312	Alcolor In Fa	Akal Metals Ine Earth Metals estilion Metals Foor Metals 4	6	Nable Gas Haligens Networks Metaloxis	7	8	9	10	11	12	13 X Sent Aluminum BA	14 20.0000 Silicon	*12.3.4.591.2.3 15 30.9736 P Phosphorus 13	2 16 32,000 Sulfur VA	VIX -1 17 33.433 Cl Chlorine VIX	1
15	+1 31.0003 K Potassium (art-art +1	*2 20 #0.870 Ca Calcium (**) #*2 *2	21 44.9999 SC Scandium (M) 341.462 +1	178 22 42.80 Ti Titanium [/0] 342 442 ~2,3,4	VB 23 50.9411 V Vanadium (Ar) 343.452 +2,3,4,5	24 91.9961 Cr Chromium [81] 345 4t1 ~2,3,6	25 54.500 Mn Manganese (M) 365.422 42,3,4,7	25 55.045 Fe Iron (M) 365.462 42,3	27 51.900 Cobalt [M] 347 462 423	VIII 228 SLIRH Nickel (4) 38 4c2 42,1	29 63.94 Cu Copper (M) 3410 411 +1,2	118 530 63.409 Zn Zinc [Ar] 3410 442 -2	4) 31 80.725 Ga Gallium (Ar) 5r10 4r2 4r1 -3	42,44 32 72.54 Ge Germanium [Ar] 3418 462 462 42,4	As As	#4342	*1,5,761 36 71.904 Br Bromine (Ar) 3410 462 465 *1,561	-
37	7 80.4676 Rb Rubidium (%) 5s1 +1	38 #7.82 Strontium (%) 5x2 42	39 83.999 Y Yttrium (M) 445 562 +3	40 91.224 Zr Zirconium (0) 402 502 -4	41 52,004 Niobium (10) 444 Sc1 +3,5	42 19.34 Mo Notybdenum [10] 445 5x1 -4	TC Internetion	Ru	Rh	Pd Pd Pellectures	Ag	Cd Cethnian	In holes	50 198,71 Sn Tin [0] 4410 562 562 +2,4	SP	Te	53 126.904     lodine  N/j 4#10 5x2 5p5 +1,5,7k1	11 
	C1 Craiter	66 107.307 Ba Barium [76] 612 -2	Lanthanide Series	72 576.49 Hafnium [26] 4714 542 612 -4	73 saused Ta Tantalum (Ne) 4714 545 612 +6	74 ND.64 W Tungsten (Xe) 4544 Sold Sold His	Ra	Os	IC Internet of the second	Pt	Au	Hg	TI Durbum 75.00	Philip	Bi	Po	At	10.00
	Fr	Ra	Actinide Series	RI	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Uut	Uuq	Uup	Uuh	Uus	U
		Lanthankles	67 x30.366 La Lanthanum (24) Set 652 =)	68 148.116 Ce Cerium [Xe] 40 Set 612 -3,4	69 148.388 Pr Praseodymium [74] 45 452 +0	60 146.24 Nd Neodymium [Ne] 44 652 +2	Pm	62 198.36 Sansarium [76] 45 612 42,0	Eu	64 197.29 Gd Gadolinium (Ne) 47 541 612 4)	Tb Tothins protect	65 1823 Dy Dysprosium [Xe] 4710 652 +0	Ho	Er	Tm Tm Durium Incense	Yb	La	
		Actinides		Th	Pa	U	Np Naphaniani	Pa	Âm	Cm		Cf	Es	Em	Md		Li	

• And let's get rid of the elements that are truly rare such as platinum, palladium, gold, silver, etc.

Ē	Group 1 LA					Peri	odic	Tabl	e of	the E	lem	ents	- Net					VI
	H Rytropen	2 ILA	No	synthe	tic, no i	adioac	tive, no	inert, n	o toxic	, no rare	e, no si	alt-form	Ing elen	14 IVA	15 VA	ogen 15 VIA	17 VIIA	H
		Be	Categoria	e at STP Liquid pories	Solid	Synthetic							6 Man B Boron	Carbon	7 100007 N Nitrogen	8 11.9994 O Oxygen VA		1.1.1
	Na	Mg Mg Magarrans	3	ection Metals In Earth Metals Poor Metals 4	5	Kan metalt Metalout	7	8	9	10	11	12	13 X.SEIS Al Aluminum BA	14 20.0055 Silicon	15 30.9276 P Phosphorus XA	S Sultan W	GI Ciliartes	1
	K	Ca Ca Cateline	21 44.9999 SC Scandium (M) 341 442	22 0.307 Ti Titanium (4) 302 402	23 50.9415 V Vanadium [N] 345 442 +2 3.4.5	24 91.9961 Cr Chromium (Ar) 345 411 (2.3.6	25 34.500 Mn Manganese (M) 365.82 52.447	25 55.00 Fe Iron [M] 365 462	27 54.903 Co Cobalt [M] 347 462	228 SLIBH Nickel (N) 198 42	29 KLM Cu Copper [Ar] 3410 411	530 63.400 Zn Zinc [M] 3410 442	4) Ga Gallium [Ar] 3110 412 4p1	42,404 32 72,54 Germanium [Ar] 3610 422 42 424	As	Se Se	Br	N NO
	Rb	38 87.82 Sr Strontium (9) 562 42	39 00.9999 Y Yttrium (0) 445 5c2 +3	40 91.224 Zr Zirconium (04 40 502	41 \$2,9964 Nb Nioblum (0) 444 Sc1 +3.5	42 M.H Mo Molybdenum (10) 445 for 1	TG bennetime	Ru	Rh	Pd Pd Pelledian	Ag	Cd	In House	50 198.79 Sn Tin [0] 4410 5n2 5p2 +2.4	Sp Sp	Te Te Telation		24 
	Ca	66 137.329 Ba Barium (%) 812 42	Lanthanide Series	72 578.49 Hf Hatnium [24] 4714 542 612	73 508.940 Ta Tantalum (26) 4714 545 612	74 103.64 W Tungsten [26] 4514 564 652 -4	Ra	0s	Ic Ic Intitue	Pt	Au	Hg	TI Dathas	Pb	Bi	Po	At	RAN
	Fr	Ra	Actinide Series	RI	Ob	Sg	Bh	H H	MIL	Ds	Rg	Cn	Uut	Uuq	Uup	Uuh	Uus	U
		Lanthankles	67 130.996 La Lanthanum (xe) 541 652 <)	68 148.116 Ce Cerium [Xe] 41 541 612 +3,4	65 146.300 Pr Praseodymium [70] 45 052 -0	60 14629 Nd Neodymium [Ne] 46 652 -)	Pm Pm	62 199.36 Sm Samarium [24] 45 652 42,3	Eu Eu fanatan	64 197.29 Gd Gadelinium (24) 47 5m 6s2 +3	Tb Trikim	65 162.5 Dy Dysprosium [Xe] 400 612 +0	Ho Ho Ridenius Jonarth	Er	Tm The Declars	Yb Yb Nachian Nonese	Lu Lu Lulium	
		Activides	Ac	Th	Pa	U	Np	Pu		Cm		GI	Es		Md	No		

- Oh and hydrogen and the rock-forming elements won't do us any good either at least that's our tentative belief.
- So we're down from 90 naturally occurring elements to 36 still a lot to work with.
- Let's ask a question: what elements have been used over the last 150 years to make magnetic materials?

		15		-^	151			~9			materiale
	Ma	jor co	nstitu	uent	s		Mir	nor co	onsti	tuents	Comments
t Magnetic Material	s										
Iron	Fe										Low carbon mild steel
Silicon Steel	Fe						Si				Si at 2.5 to 6%
Nickel-Iron	Fe	Ni									Ni at 35 to 85%
Moly Permalloy	Ni	Fe					Mo				Ni at 79%, Mo at 4%, bal. Fe
Iron-Cobalt	Fe	Co					V				23 to 52% Co
Soft Ferrite	Fe	Mn	Ni	Zn			0				
Metallic Glasses	Fe	Co	Ni				в	Si	Р		Amorphous and nanocrystalline
r <b>manent Magnets</b> Co-Steels	Fe	Co									
r <b>manent Magnets</b> Co-Steels Alnico	Fe Fe	Co Ni	Co	Al	Cu		Ti	Si			
r <b>manent Magnets</b> Co-Steels Alnico Platinum Cobalt	Fe Fe Pt	Co Ni Co	Co	Al	Cu		Ti	Si			
r <b>manent Magnets</b> Co-Steels Alnico Platinum Cobalt Hard Ferrites	Fe Fe Pt Fe	Co Ni Co Sr	Co	Al	Cu		Ti	Si			Oxygen dilutes; Ba no longer used
r <b>manent Magnets</b> Co-Steels Alnico Platinum Cobalt Hard Ferrites SmCo	Fe Fe Pt Fe Co	Co Ni Co Sr Sm	Co (Gd)	Al	Cu Cu	Zr	Ti	Si			Oxygen dilutes; Ba no longer used
rmanent Magnets Co-Steels Alnico Platinum Cobalt Hard Ferrites SmCo Neodymium-iron-boron	Fe Fe Pt Fe Co	Co Ni Co Sr Sm Nd	Co (Gd) Dy	Al Fe (Y)	Cu Cu B	Zr Co	Ti Cu	Si Ga	Al	Nb	Oxygen dilutes; Ba no longer used
rmanent Magnets Co-Steels Alnico Platinum Cobalt Hard Ferrites SmCo Neodymium-iron-boron Cerium-iron-boron	Fe Fe Pt Fe Co Fe Fe	Co Ni Co Sr Sm Nd Nd	Co (Gd) Dy Ce	Al Fe (Y) B	Cu Cu B	Zr Co	Ti Cu	Si Ga	Al	Nb	Oxygen dilutes; Ba no longer used Limited use in bonded magnets
rmanent Magnets Co-Steels Alnico Platinum Cobalt Hard Ferrites SmCo Neodymium-iron-boron Cerium-iron-boron SmFeN	Fe Fe Pt Fe Fe Fe Fe	Co Ni Co Sr Sm Nd Nd Sm	Co (Gd) Dy Ce N	Al Fe (Y) B	Cu Cu B	Zr Co	Ti Cu	Si Ga	Al	Nb	Oxygen dilutes; Ba no longer used Limited use in bonded magnets Nitrogen is interstitial; stability is
rmanent Magnets Co-Steels Alnico Platinum Cobalt Hard Ferrites SmCo Neodymium-iron-boron Cerium-iron-boron SmFeN MnBi	Fe Fe Co Fe Fe Fe Mn	Co Ni Co Sr Sm Nd Nd Sm Bi	Co (Gd) Dy Ce N	Al Fe (Y) B	Cu Cu B	Zr Co	Ti Cu	Si Ga	AI	Nb	Oxygen dilutes; Ba no longer used Limited use in bonded magnets Nitrogen is interstitial; stability is Never commercialized

- This list contains most (though not all) common magnetic materials and the elements used to make them.
- Take a good look and then move to the next slide showing them on the periodic table.

1	Group 1 LA	1			Elen	nents	use	d in I	Exis	ting N	Magr	netic	Mate	erials	5			18 VIII
	H	2 ILA		02273									13 IIIA	14 IVA	15 VA	16 VIA	17 VIIA	H
ľ		Be	Cate	gories	Solid	Synthetic Name Gas							6 Man	Carbon	Nitrogen	S 13.9994 Oxygen Via	F	N
Ī	Na	Mg	3	ancilion Metalt me Eurih Metalt Poor Metalt 4	5	Non-metals Metaloots	,	8	9	10	11	12	13 X.SHI Al Aluminum	14 20.000 Silicon	Phosphorus	S Sullar	GI	A
	K K Presson	Ca	SC SC	1VB 22 0.00 Ti Titanium [M] 302 462	VB 23 50.9411 V Vanadium (Nr) 343-412	VIB 24 91.9961 Cr Chromium [24] 345 4:1	VIIB 25 54.500 Mn Manganese (M) 365.462	VIII 25 55.645 Fe Iron (M) 365.462	VIII 27 58.900 Cobalt [44] 347 442	VIII 228 SLARM Nickel [Ar] 38 4e2	18 29 61.14 Copper [Ar] 3410 441	118 6 30 63.409 Zn Zinc [Ar] 5410 442	4) 31 81.722 Gallium (Ar) 3r10 452 4r	Ge	ASS ASS ASS ASS ASS ASS ASS ASS ASS ASS	Se Se Second	Br Area Br Branies	K
	Rb	38 87.62 Sr Strontium (0) 562 42	39 00.900 Y Yttrium (0) 441 562 *3	240 91.224 Zr Zirconium (v) 442 562	41 52,564 Nb Nioblum (0) 444 Set +3.5	42 W.H Mo Molybdenum (V) 445 for 4	Te	Ru	Rh	Pd	Ag	Cd Cethnian Sectorian		Sn The	Sb Sb	Te Te Telastan	li janina li janina li janina	21 X
	Ca	66 137.329 Ba Barium (%) 852 42	Lanthanide Series	72 578.49 Hf Hatnium [74] 474 542 612 +4	73 588.940 Ta Tantalum (Ne) 4714.545 (s.2 +6	74 103.04 W Tungsten (26) 4514 554 652 -6	Re	0s	IC IC Intition Internet	28 million Platinum (24) 414 Schut <2,4	Au	Hg	TI Dathan 1924	Pb	Bi	Po	At	R
	Fr	Ra	Actinide Series	RI	Db Hereit	Sg	Bh	H 1	ALL MIL Sectors from	Ds	Rg	Cn Name	Uut	Uuq	Uup	Uuh	Uus	Uü
		Lanthankies	67 tauser La Lanthanum (xe) Set 652 +7	6 68 148.116 Ce Cerium [Xe] 41 541 612 -3,4	69 H8.98 Pr Praseodymium [76] 45 652 -0	60 14624 Nd Neodymium [24] 45 652 -3	Pm	62 198.36 Sm Samakan (24) 45 612 -2,3	Eu Eu faceptum registran	64 197.29 Gd Badolinium [Se] 47 541 482 4)	Tb Trikien (states)	65 162.5 Dy Dysprosium [Xe] 410 812 -0	Ho Ho Natasian Internet	Er	Tm	Yb	Lu	
		Activides	Ac	Th	Pa	U U U U	Np Nephanian	Pu	Âm	Cm		Cf	Es	Fm	Md	No	Lr	

- These elements are, with three exceptions, the same elements we selected by paring down the periodic table.
- The exceptions:
- 1) platinum-cobalt was the first high performance magnet. It was used to make watch drive motor magnets whose very small size compensated for the high material cost.
- 2) Germanium and Tin have not been used except as very minor additives, at least to my knowledge, in commercial magnets, but like aluminum and gallium might make suitable modifying constituents to assist sintering or phase formation.



- Existing research projects is split here into two categories.
- The first could be called "Variation on a Theme" as it represents an extension of research on materials that have been previously examined and it is shown in the list on this slide.
- However, there are several differences between what took place "then" and what is being pursued "now".
- One difference is that current analytic capabilities are superior to what existed even two or three decades ago.
- Secondly, we now have techniques to form these materials with a refined structure at micro- and nano-scales.
- Research is focused on materials that exhibit ferromagnetic properties either naturally or when combined with alloying elements with a focus on the structure.



- We started with 90 naturally occurring elements and ended with 36 promising ones.
- The "bottoms-up" approach is to take the 36 remaining elements from our experiment with the periodic table and to combine them using computer algorithms to forecast the potential for generating a magnetic moment.
- Then the list of most promising alloys must be produced in the lab and evaluated.
- One of the more significant hurdles is to make a nano-structured material fully dense and to do so in a scalable, economic manner.
- The beneficial properties of magnetic materials are due in part to either crystal shape anisotropy (e.g. alnico) or magneto-crystalline anisotropy (e.g. ferrite, SmCo and Neo).
- In either case, during manufacture the magnetic domains must be mutually aligned "coparallel" to obtain maximum properties.
- Simultaneous densification and alignment of nano-particulates has been a difficult problem awaiting solution.



- Heusler alloys have interesting crystalline structures.
- They were first identified as a family of materials in 1905 and have found recent revival in spintronics.
- A Heusler alloy is a ferromagnetic metal alloy based on a Heusler phase. Heusler phases are intermetallics with particular composition and face-centered cubic crystal structure. They are ferromagnetic—even though the constituting elements are not—as a result of the <u>double</u>-exchange mechanism between neighboring magnetic ions. The latter are usually manganese ions, which sit at the body centers of the cubic structure and carry most of the magnetic moment of the alloy.

(Wikipedia)



- In current research, in addition to the importance of structure is the importance of thermal processing in the development of optimal microstructure "phase equilibrium 101".
- With the exception of ceramic (hard ferrite) magnets, magnetic alloys are just that alloys.
- Therefore, thermal treatments to form the stable and desirable phase structure is not only preferable but also very common.
- In the chart at the left, Strnat shows the development of the hysteresis loop of SmCo 2:17 during its thermal treatment.
- In the chart to the right, we see the improvement of magnet properties of alnico due to thermal processing in the presence of an aligning magnetic field.



- In another example, alnico is solution treated at high temperature (about 1230 °C) followed by a conditioning treatment effected by controlled cooling from the solution treatment temperature or by isothermal treatment of the magnets anisotropic magnets are treated in a field during spinodal decomposition at ~820 °C.
- The third and final treatment is called a "draw" or "coercive aging treatment" to obtain maximum coercivity and optimal loop shape.
- We might say that the right composition provides the opportunity and the right thermal treatment creates the right phase structure.



- Problems with pricing and supply of permanent magnets include both logistic and technical issues.
- Technical issues are resolved by R&D for new, improved materials which strategy is suitable as a long-term plan but is highly unlikely to have any short-term influence.

